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- Polymer mixture having an aromatic polycarbonate and an aromatic polyester.
- The invention relates to polymer mixtures which comprise an aromatic polycarbonate and an aromatic polyester. The polymer mixtures are stabilised by means of phosphorus-containing stabilisers known per se and elementary sulphur.

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Polymer mixture having an aromatic polycarbonate and an aromatic polyester.

The invention relates to a polymer mixture which comprises the following constituents:

- A. 1-98.99% by weight of an aromatic polycarbonate,
- B. 98.99-1% by weight of an aromatic polyester, and
- C. 0.01-2% by weight of a stabiliser.

It is known, for example, from GB-A-1466154, that polymer mixtures which comprise an aromatic polycarbonate and an aromatic polyester show a certain instability.

A transesterification reaction probably occurs in such mixtures.

It is proposed in GB-A-1466154 to suppress the said transesterification by the addition of a phosphoruscontaining compound, for example, triphenyl phosphite, phosphorous acid.

The invention is based on the discovery that a further improvement of the stability can be obtained.

The polymer mixture according to the invention is characterised in that as a stabiliser it comprises a phosphorus containing stabiliser known per se and elementary sulphur, in a quantity from 0.0001-1% by weight calculated with respect to the sum of the parts by weight of A + B + C.

The use of elementary sulphur and a phosphorus-containing stabiliser known per se results in a slightly improved stability compared with polymer mixtures which comprise only a phosphorus-containing stabiliser.

The known phosphorus-containing stabilisers have the additional disadvantage of showing only a slight stabilising activity in polymer mixtures which comprise one or more pigments, for example, ultramarine blue and titanium dioxide. It has been found quite unexpectedly that the addition of a small quantity of elementary sulphur removes this disadvantage.

The invention also relates to articles formed from the polymer mixture according to the invention.

The polymer mixture according to the invention comprises at any rate the following constituents in the quantities indicated hereinbefore:

- A. aromatic polycarbonate,
- B. aromatic polyester.

C. a phosphorus-containing stabiliser and elementary sulphur.

The polymer mixture according to the invention may moreover comprise one or more of the following constituents:

- D. one or more pigments
- E. conventional additives.

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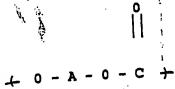
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A. Aromatic polycarbonate

Aromatic polycarbonates are materials known per se. They are generally prepared by reacting a dihydric phenol compound with a carbonate precursor, for example, phosgene, a halogen formiate or a carbonate ester. Aromatic polycarbonates are polymers which comprise units of the formula



wherein A is a bivalent aromatic radical derived from the dihydric phenol which is used in the preparation of the polymer. As dihydric phenois may be used in the preparation of the aromatic polycarbonates mononuclear or polynuclear aromatic compounds which comprise two hydroxy radicals which may each be directly bonded to a carbon atom of an aromatic nucleus.

50 2,2-bis-(4-hydroxyphenyi)propane; hydroquinone; resorcinol; 2,2-bis-(4-hydroxyphenyi)pentane; 2,4'-(dihydroxydiphenyl)methane: bis(2-hydroxypnenyl)methane; bis-(4-hydroxyphenyl)methane; bis-(4-hy 5-nitrophenyl)methane; 1,1-bis-(4-hydroxyphenyl)ethane; 3,3-bis(4-hydroxyphenyl)pentane; 2,2-dihydroxyphenyl; 2,8-dihydroxynaphthalene; bis-(4-hydroxydiphenyl)sulphone; bis-(3,5-diethyl-4-hydroxyphenyl)sulphone; 2,2-bis-(3,5-dimethyl-4-hydroxyphenyl)propane; 2,4-di-hydroxyphenyl sulphone; (5-chloro-2,4dihydroxy-diphenyl)sulphone; bis-(4-hydroxyphenyl)diphenyl sulphone; 4.4 dihydroxydiphenyl ether; 4.4 dihydroxydiphenyl

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dihydroxy-3,3/-dichlorodiphenyl ether; 4,4'-dihydroxy-2,5-dihy-droxydiphenyl ether.

Other likewise suitable dihydric phenols are disclosed in US-A-2,999,835; 3,038,365; 3,334,154, and

The aromatic polycarbonates may be prepared according to methods known per se: for example, by 4,131,575. reacting a dihydric phenol with a carbonate precursor, for example, phosgene. For this purpose, reference may be made to the United States Patent Specifications mentioned hereinbefore and to US-A-4,018,750 and 4,123,426. They may also be prepared by a transesterification as described in US-A-3,153,008.

The branched polycarbonates known per se as described, for example, in US-A-4,001,184 are also

Suitable aromatic polycarbonates are also the so-called polyester carbonates which are obtained by suitable. carrying out the polymerisation reaction in the presence of an ester precursor, for example, a difunctional carboxylic acid, for example, terephthalic acid or an ester-forming derivative thereof. These polyester carbonates have ester compounds and carbonate compounds in the polymeric chain. Polyester carbonates are described, for example, in US-A-3,169,121.

It is also possible in the polymer mixtures according to use a mixture of various polycarbonates, for example, as mentioned hereinbefore, as an aromatic polycarbonate.

B. Aromatic polyester

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Any aromatic polyester known per se may be used in the polymer mixtures according to the invention.

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The polyalkylene terephthalates are particularly sultable polyesters. Polyalkylene terephthalates are compounds known per se. They may be described as glycol esters of terephthalic acid. They may be prepared, for example, by alcoholysis of esters of terephthalic acid with a glycol succeeded by a polymerisation reaction, by heating glycol compounds with free acids or derivatives thereof. The glycol part of the polyalkylene terephthalates may comprise from 2 to 10 carbon atoms; it preferably comprises from 2 to 4 carbon atoms, in the form of linear alkylene chains.

Polyesters derived from ethylene glycol or butane-1,4-diol and terephthalic acid are preferably used. It is also possible to use copolyesters in which a part of the said glycol and/or of the terephthalic acid is replaced by another glycol and/or aromatic carboxylic acid. Generally, not more than 30 mcl.%, preferably not more than 10 mol.%, of the glycol and/or terephthalic acid are replaced by other comonomers in such

It is also possible to use so-called block copolyesters as a polyalkylene terephthalate. The said block copolyesters are prepared, for example, by converting a polybutylene terephthalate with reactive terminal copolyesters. groups with a reactive polyester or copolyester in the presence of a transesterification catalyst.

It is further possible to incorporate a branching agent in the polyalkylene terephthalate, for example, a glycol having three or more hydroxyl groups or a trifunctional or multifunctional carboxylic acid.

It is also possible to use a mixture of various polyesters as the polyester.

C. Stabiliser

The polymer mixture according to the invention comprises as a stabiliser one or more phosporuscontaining compounds and elementary sulphur known per se for this purpose. For the known phosphoruscontaining stabilisers may be referred, for example, to GB-A-1486154 and GB-A-1569296. All phosphorus containing compounds mentioned in these patent papers may be used in the polymer mixtures according to

The polymer mixtures according to the invention comprise as a stabiliser in addition 0.0001-1% by weight of elementary sulphur, calculated with respect to the sum of the parts by weight of A + B + C.

The stabilisers as used in the polymer mixtures according to the invention have the advantage of not D. Pigments

Therefore, the polymer mixtures according to the invention may comprise one or more pigments, for losing or hardly losing their activity in the presence of pigments. example, ultramarine blue and titanium dioxide.

E. Conventional additives

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The polymer mixture according to the invention may comprise as additives, for example, polyolefins, mould-release agents, agents to improve the flame retarding properties, further stabilisers, for example, thermal stabilisers, dyes, fillers and reinforcing fillers, for example, glass fibres.

More in particular, the polymer mixture according to the invention may also comprise a combination of various of the additives mentioned hereinbefore.

The polymer mixture according to the invention preferably comprises a weight ratio of the quantity of polycarbonate (A) to aromatic polyester (B) between 1:8 and 8:1.

The polymer mixture according to the invention can be obtained according to conventional methods of preparing polymer mixtures, for example, by melt extrusion. The elementary sulphur may be added, for example, by first pre-mixing it in the dry state with one or more of the remaining constituents.

The invention will now be described in greater detail with reference to the ensuing specific examples:

Comparative examples A. B. C and D. examples I. II. III and IV.

Seven polymer mixtures were prepared having a composition as indicated in Table I hereinafter. The

Polybutylene terephthalate having an intrinsic viscosity of 1.10 dl/g measured in a 60/40 mixture constituents used were the following: PBT: of phenol and

Polycarbonate derived from bisphenol A and phosgene, having an intrinsic viscosity of 53.0 ml/g. 1,1,2,2,-tetrachloroethane at 25°C. measured in methylene chloride (CH_{2Cl₂1} at 25 °C.

Phosphite, namely a mixture of various aromatic and aliphatic phosphites. FOS:

Elementary sulphur. S:

Ultramarine blue: pigment 29 Pigment:

The indicated constituents were extruded collectively on a double-blade extruder at a melting tempera-TiO2:

ture of approximately 280°C and a speed of 200 rpm.

The stability against transesterification was determined as follows. Test pieces for determining the Vicar-B value according to DIN 53460 were injection-moulded from the polymer mixtures according to all the examples. The test pieces were injection-moulded under extra heavy conditions (285°C, 6 minutes). In a transesterification reaction, products are formed which lead to a reduction of the Vicat value. A lower Vicat value thus indicates a transesterification: the higher the Vicat, the stabler is the polymer mixture.

The resulting test pieces were also evaluated visually with regard to surface defects (so-called "splay").

Such surface defects often occur as a result of transesterification reactions.

The results found are recorded in Table I.

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TABLE I

Example	A	В	С	D	1	11	111
Composition (par	ts by w	reight)					43
o PBT o PC o FOS o S o Pigment o TiOz	45 55 -	43 53 - - 3.5 0.5	45 55 0.2 -	43 55 0.2 - 3.5 0.5	45 55 0.2 0.0035	45 55 0.2 0.0175	53 0.2 0.0035 3.5 0.5
Properties	<u> </u>			1	127	127	116
o Vicat B (*C) o Splay*	80 +	80	124	97.	. (27	: -	.•

⁻ no or slight splay; + strong splay

It may be seen from the results of Table I that the polymer mixtures according to the invention (examples I, II, and III) generally have an equal to slightly better stability than the comparative polymer

It is striking in particular that the polymer mixture according to Example D has a much worse stability mixtures A, B, C and D. than the corresponding example III according to the invention. The known phosphite stabiliser obviously loses its activity in the presence of pigments. The stabiliser according to the invention does not exhibit this disadvantage.

Claims

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- 1. A polymer mixture which comprises the following constituents:
- A. 1-98.99% by weight of an aromatic polycarbonate.
- B. 98.99-1% by weight of an aromatic polyester, and
- characterised in that the polymer mixture comprises as a stabiliser a phosphorus-containing stabiliser known per se and elementary sulphur, in a quantity of 0.0001-1% by weight calculated with respect to the
 - 2. A polymer mixture as claimed in Claim 1, characterised the polymer mixture moreover comprises a sum of the parts by weight of A + B + C.
 - 3. A polymer mixture as claimed in Claim 2, characterised in that the polymer mixture comprises pigment
 - 4. A polymer mixture as claimed in Claim 1, characterised in that the polymer mixture comprises in ultramarine blue as a pigment. addition one or more conventional additives.
 - 5. Articles formed from the polymer mixture as claimed in Claim 1.

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EUROPEAN SEARCH REPORT

Application Number

EP 88 20 0911

1	DOCUMENTS CONSIDER	ED TO BE KELE	Releva		ASSIFICAT	TON OF THE
ategory	Citation of document with indication of relevant passages	n, where appropriate,	to clair	n AF	PLICATIO	N (Int. CL4)
A	FR-A-2 279 816 (BAYER) * Claims 1-6 *		1-5	0	08 K	
A	US-A-3 591 550 (NITSCH * Claims 1,2 *	MANN et al.)	1-5		08 L	6//02
A	US-A-4 555 540 (R.W. A	WAKIAN)	1-5		,	
A	US-A-4 088 709 (SEYMOL * Claim 1 *	JR et al.)		i .		
Ä.	US-A-4 560 722 (J.A. * Claims 1-16 *	TYRELL)	1-5			
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٤	THE HAGUE	22-09-19			he invention	
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